

Capture of Slow Antiprotons by Atoms, Molecules, and Ions

James S. Cohen

*Theoretical Division
Los Alamos National Laboratory*

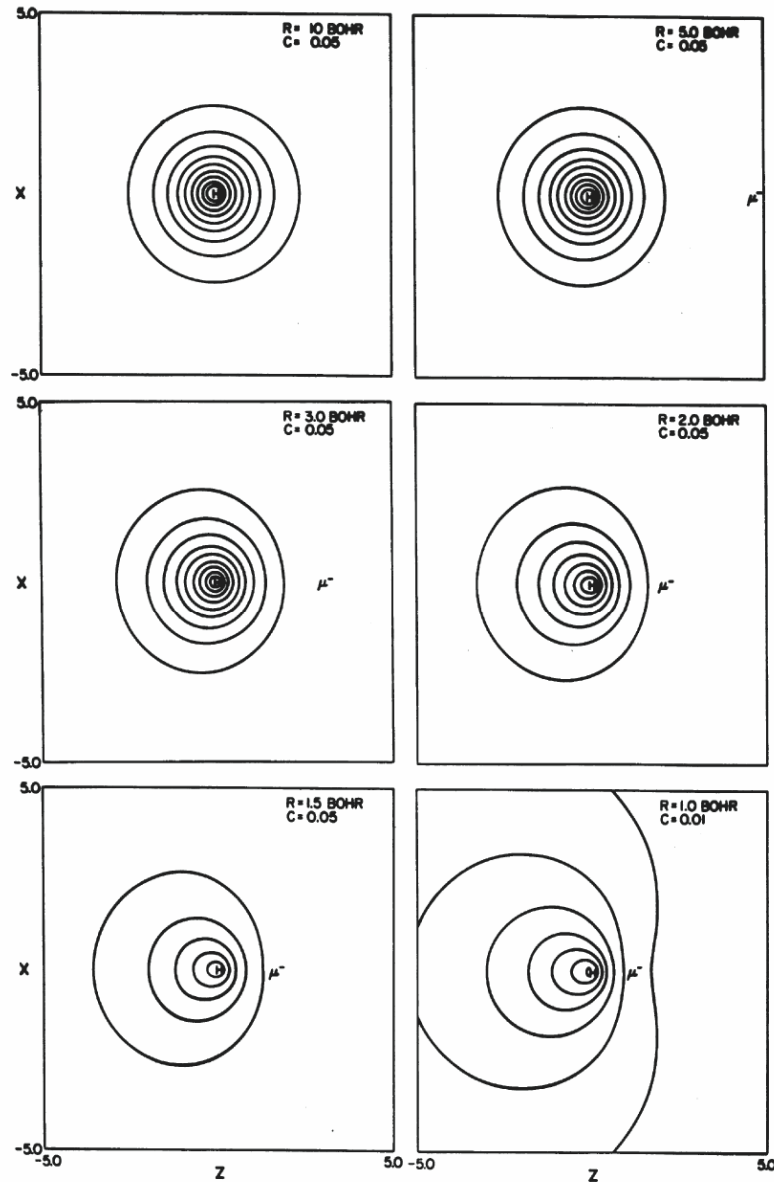
Workshop on Physics with Ultra-Slow Antiproton Beams
RIKEN, March 14–16, 2005



What do we think we know and what can be verified using the new capability with slow antiproton beams?

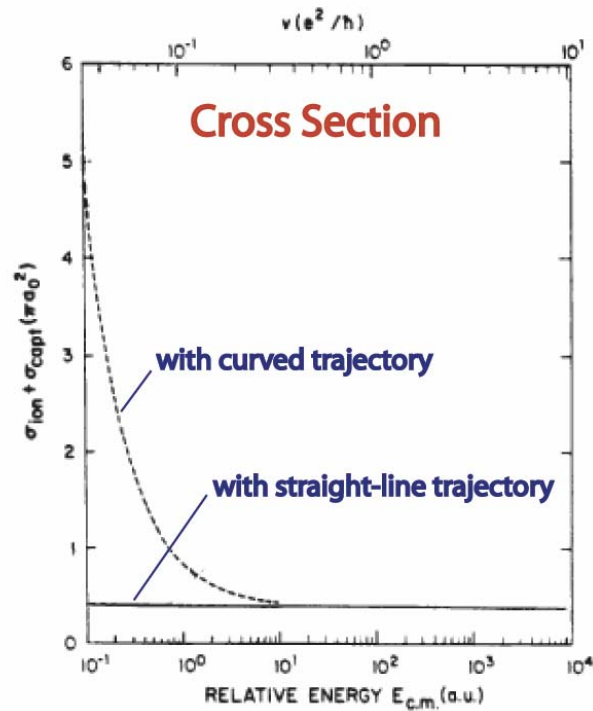
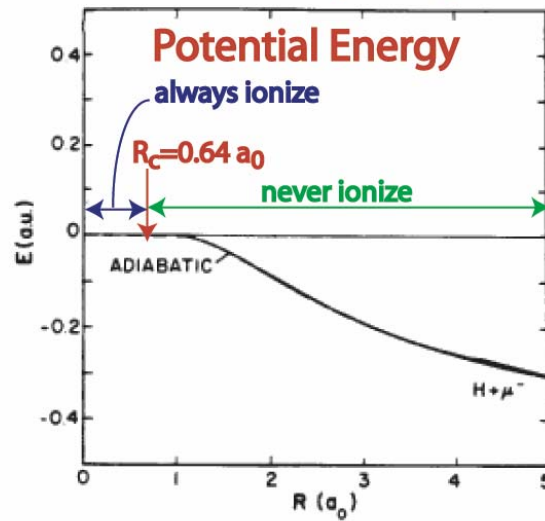
Recent review (mainly theoretical): J.S. Cohen, *Rep. Prog. Phys.* **67**, 1769-1819 (2004).

Adiabatic μ^- (or \bar{p}) + H



"Adiabatic ionization" occurs at $R < R_c \approx 0.64 a_0$
[Wightman, 1950; Fermi&Teller, 1947]

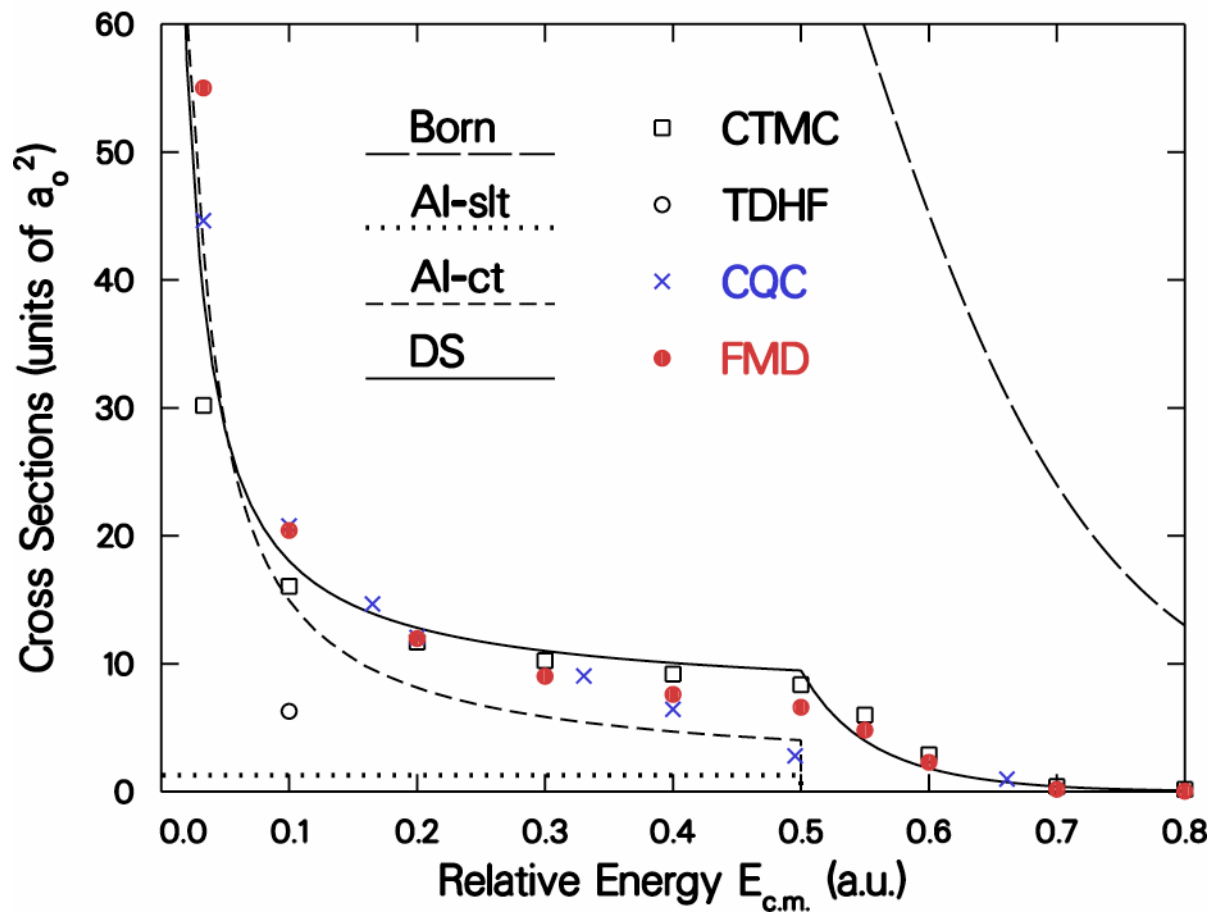
ADIABATIC IONIZATION

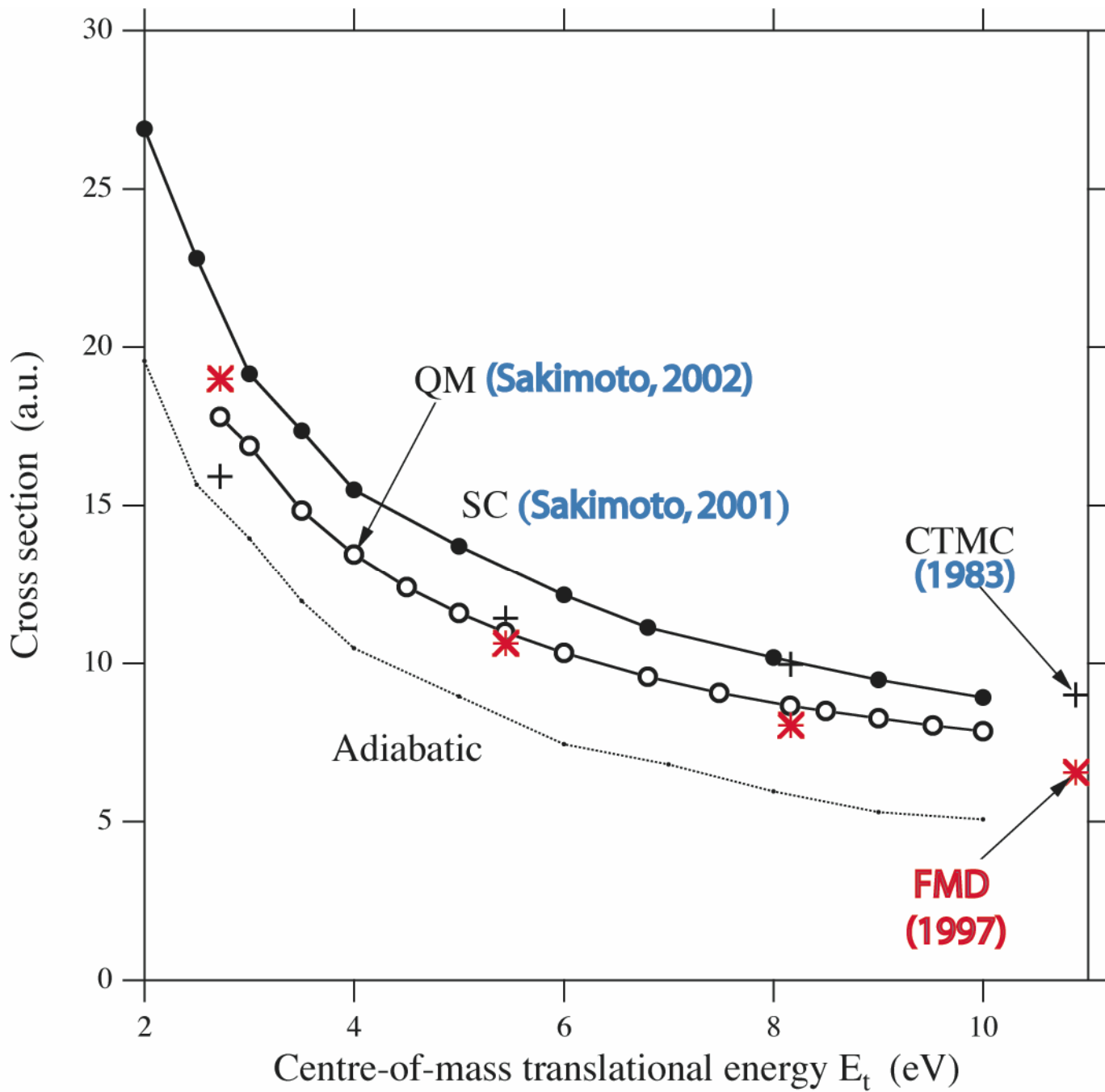


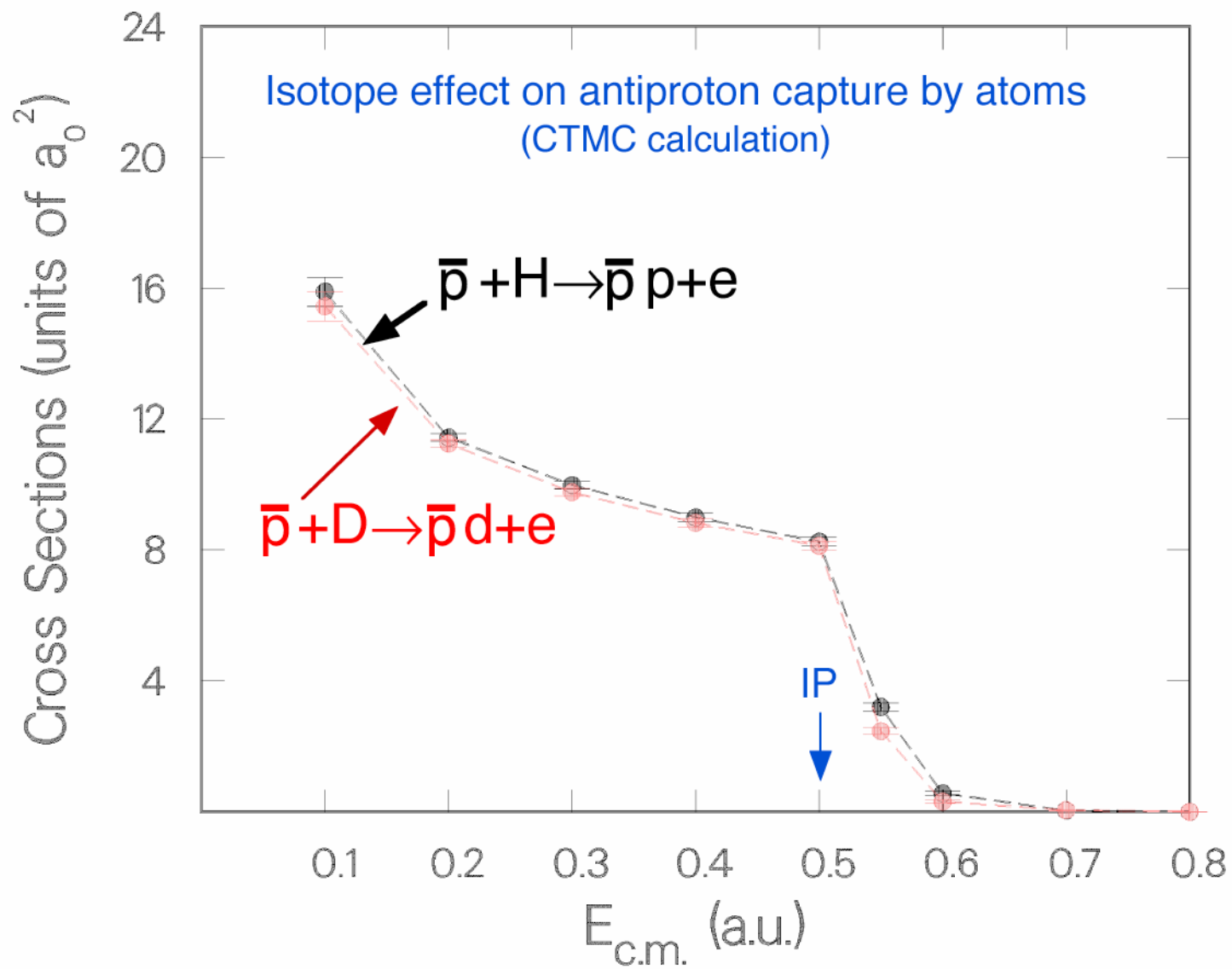
Theoretical methods for X^- capture by the H atom

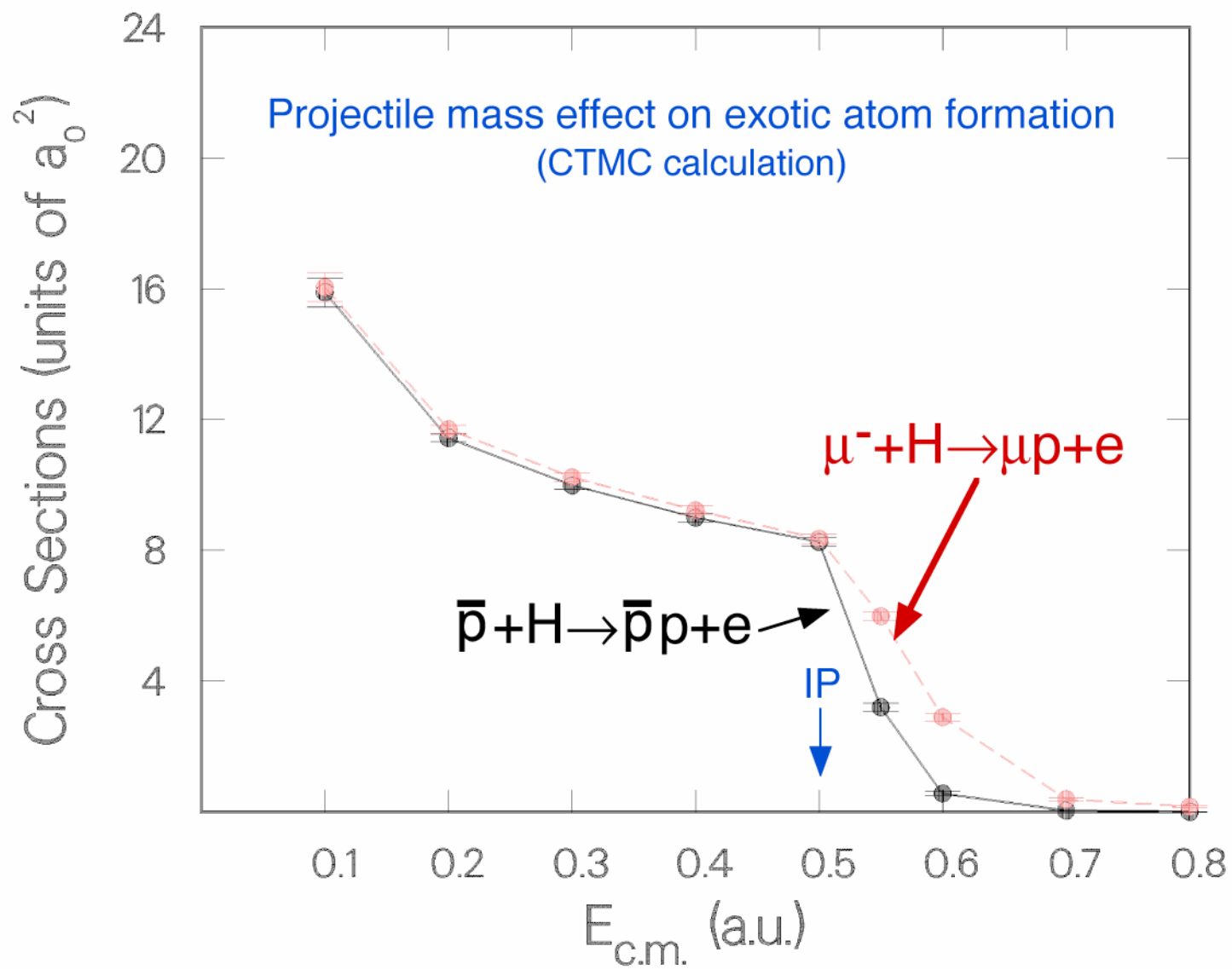
1. Adiabatic ionization (AI)
[Wightman(1950)]
2. Born / Coulomb-Born (BA, DWA)
[de Borde(1954) ... Korenman & Rogovaya(1980)]
3. Diabatic states (DS)
[Cohen, Martin & Wadt(1981)]
4. Classical-trajectory Monte Carlo (CTMC)
[Cohen(1983)]
5. Time-dependent Hartree-Fock (TDHF) with μ wave packet
[Garcia, Kwong & Cohen(1987)]
6. Classical-quantal coupling (CQC)
[Kwong, Garcia & Cohen(1989)]
7. Close coupling (CC)
[Boukour(1996)]
8. Fermion molecular dynamics (FMD)
[Cohen(1997, 1998)]
9. Perturbed stationary states (PSS)
[Ohtsuki(1999)]
10. Time-dependent semiclassical(trajjectory) Schrödinger eq. (SC)
[Sakimoto(2001)]
11. Time-dependent wavepacket Schrödinger equation (QM)
[Sakimoto(2002)]

Calculations of $\mu^- + \text{H}$ Capture









The Quasiclassical Method

The **F**ermion **M**olecular **D**ynamics (**FMD**) method is an extension of the **C**lassical-**T**rajectory **M**onte **C**arlo (**CTMC**) method to multi-electron atoms and molecules, obtained by introducing pseudo-potentials to represent quantum-mechanical effects.

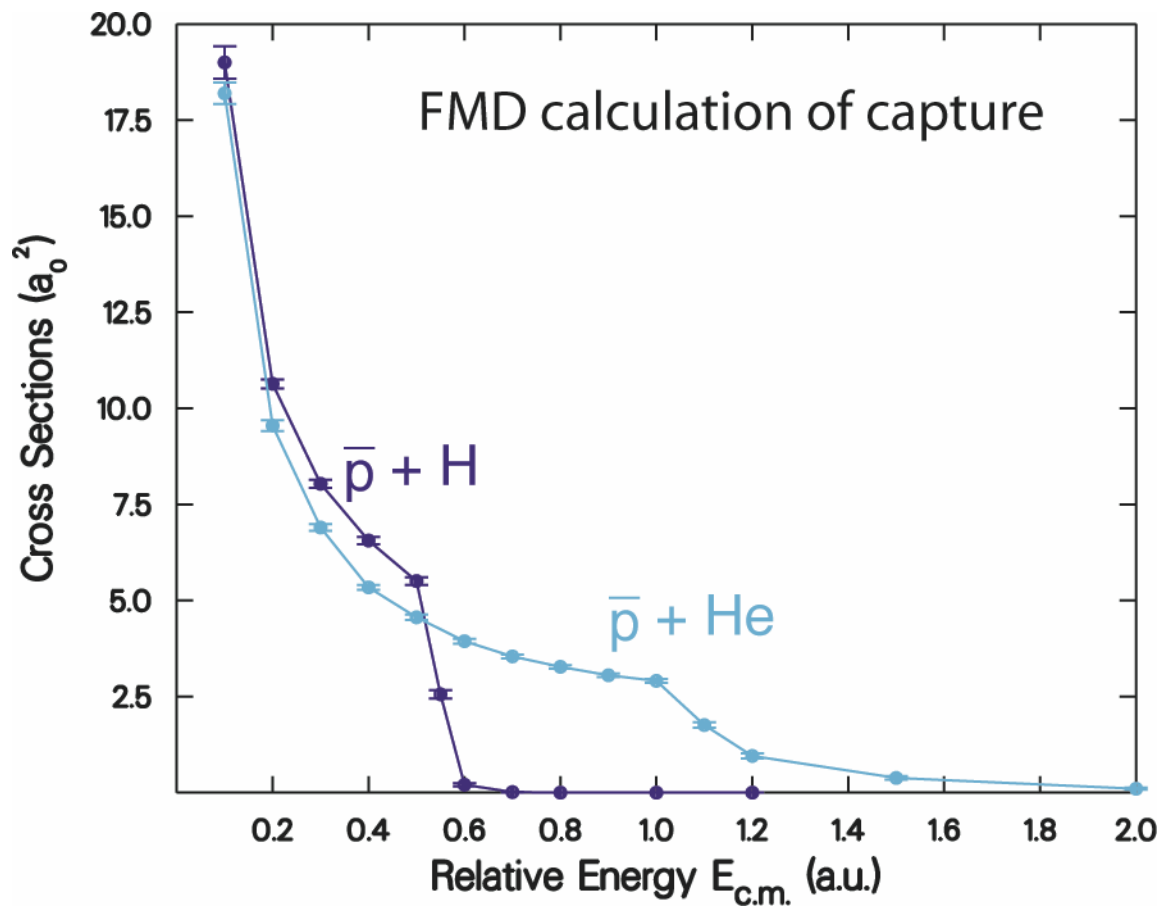
Advantages for capture of heavy ($m \gg m_e$) negative particles:

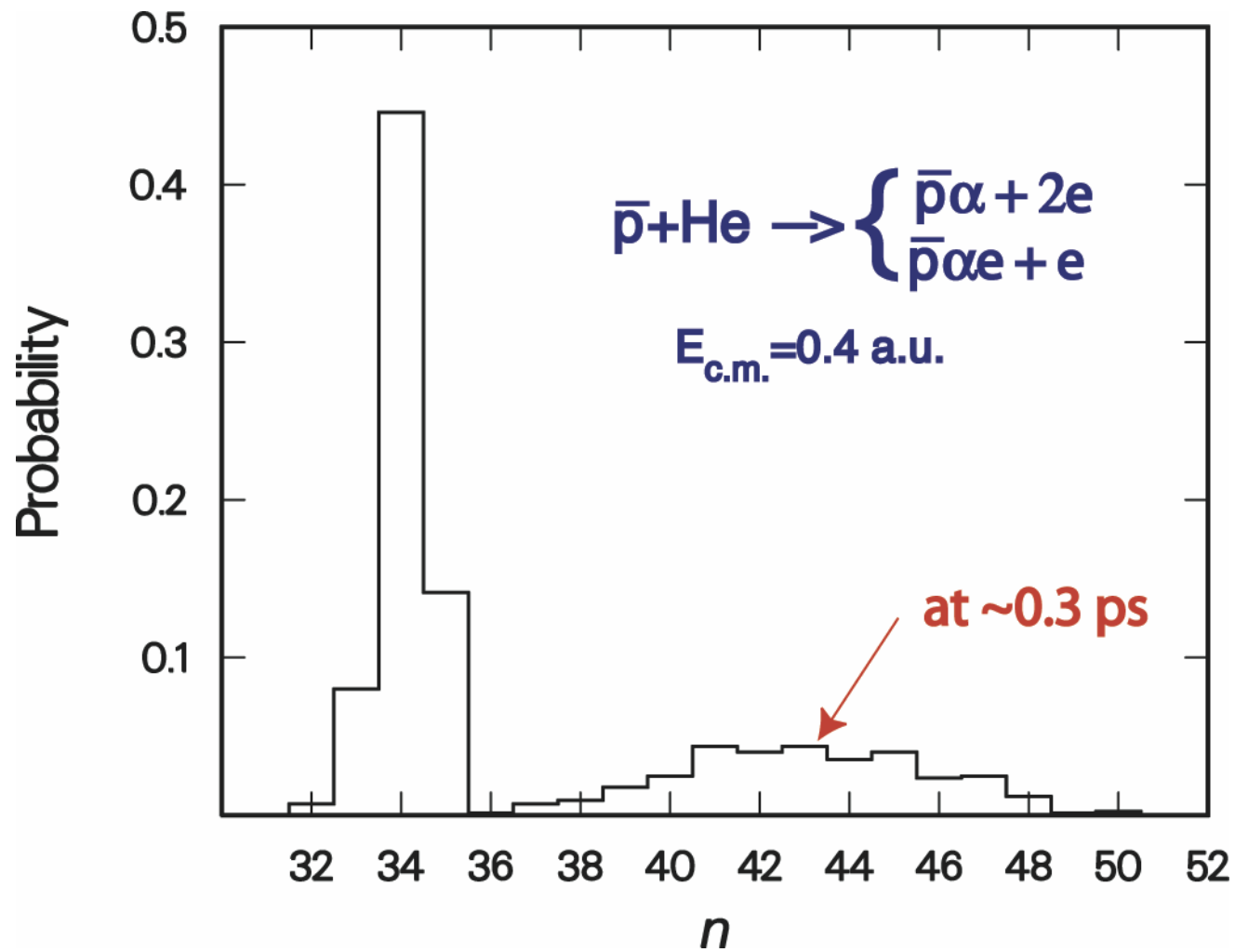
- The **classical (Hamilton) equations of motion** are easier to solve than the Schrödinger equation.
- **Full dynamics** is done for all particles. Thus correlation is fully treated and all rearrangement channels (including breakup) are treated on equal footing.
- The **electronic continuum** occurs naturally.
- The **large number of intermediate and final states** presents no difficulty. Capture is into high n, l orbitals, so the **correspondence principle** applies.

Possible weaknesses:

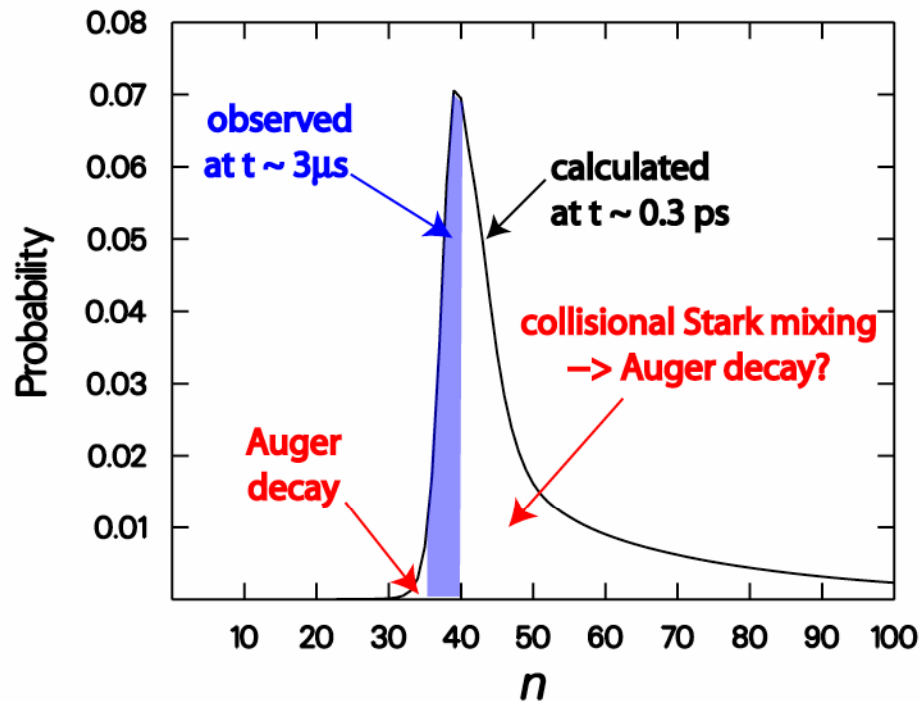
- **Correlation** may be too strong.
- May not be adequate when the **molecular vibrational–rotational spacings** are large (as in H_2).
- **Not valid for light reactants** like electrons and positrons.

Where corroboration is available, the FMD method seems to be rather accurate, but, at the very least, it is useful to obtain an **indication of important features**, so requirements for experiments and future quantum-mechanical methods can be gauged.





Metastable $\bar{p}\alpha\epsilon$

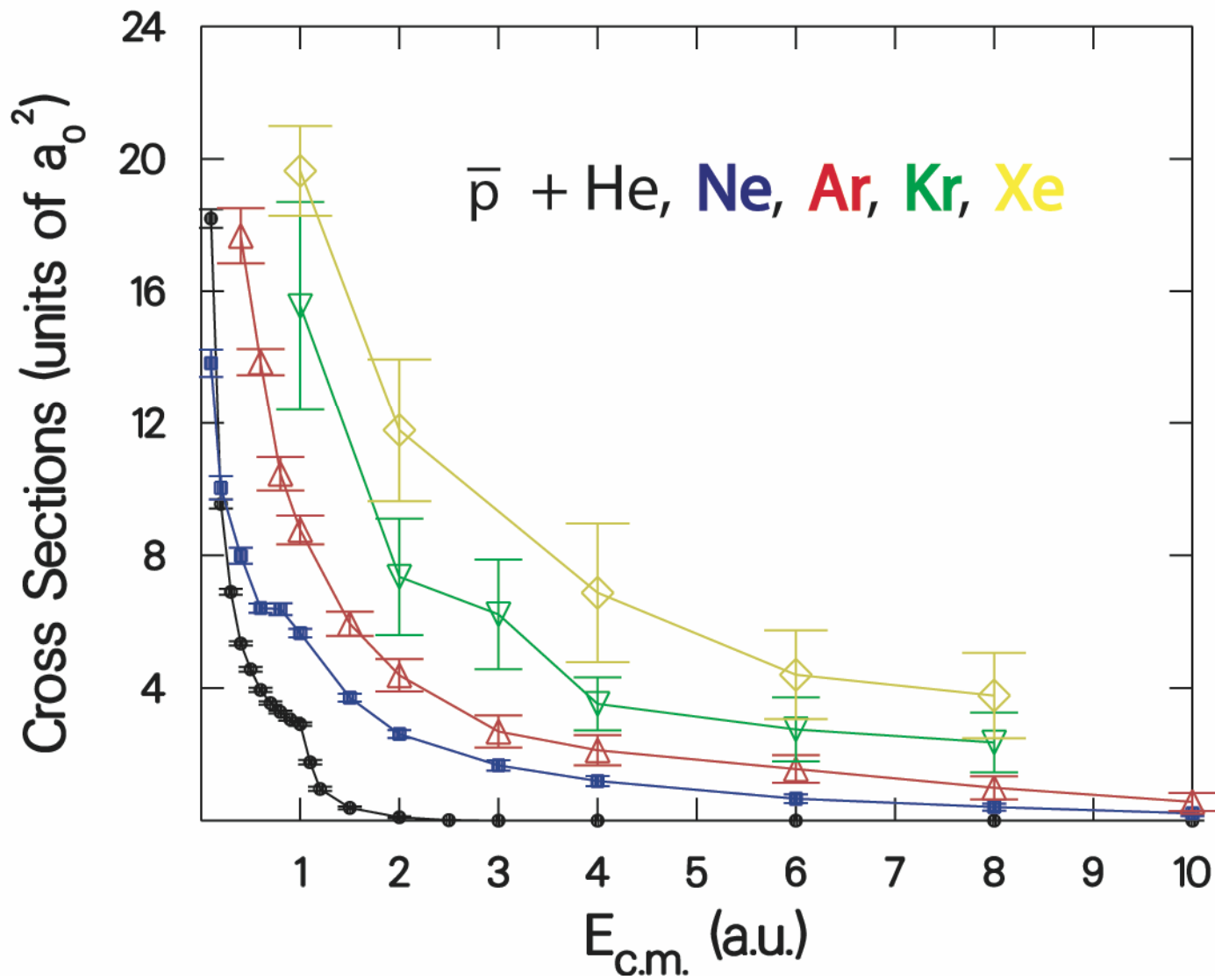


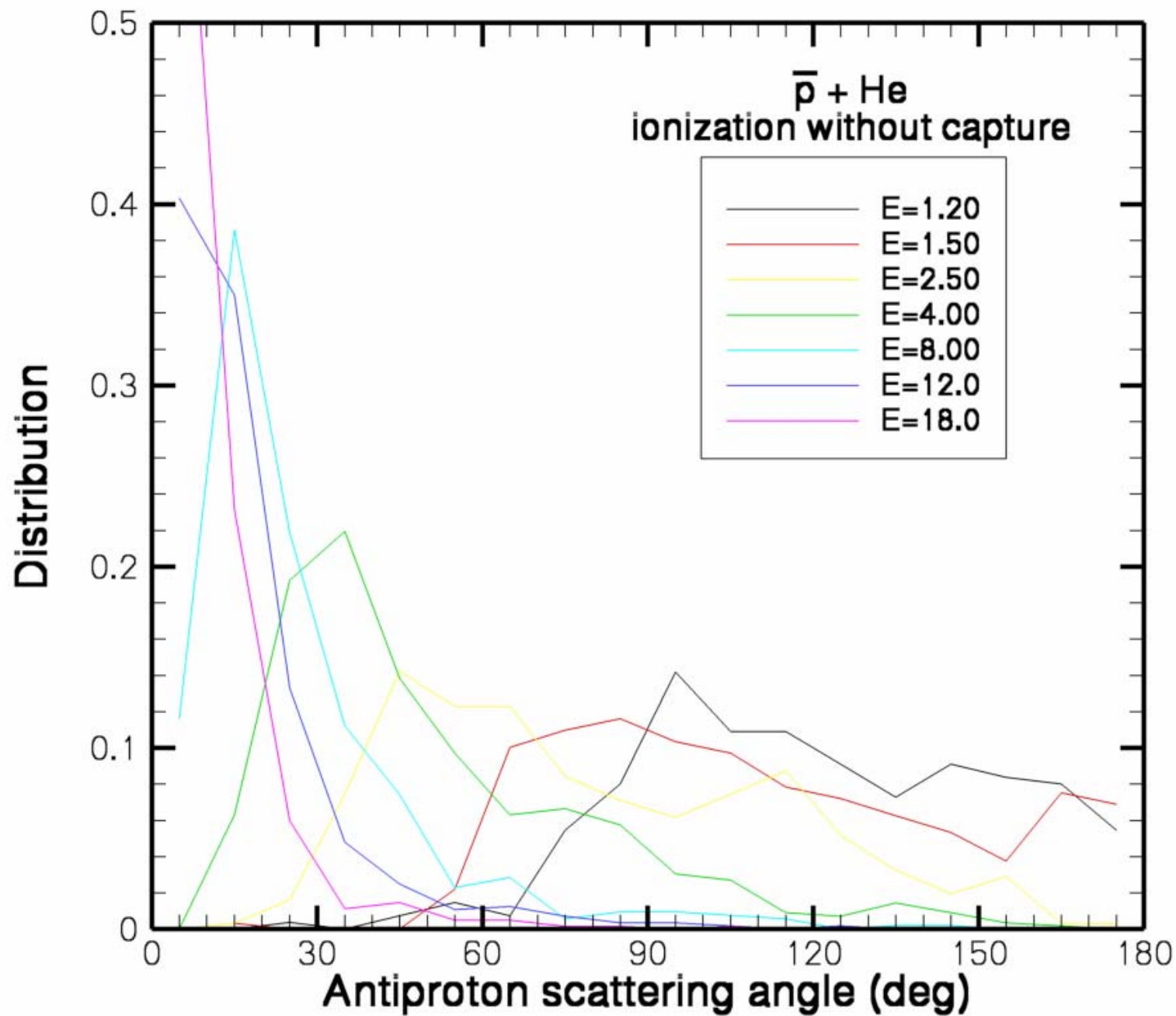
States can **Auger decay** if

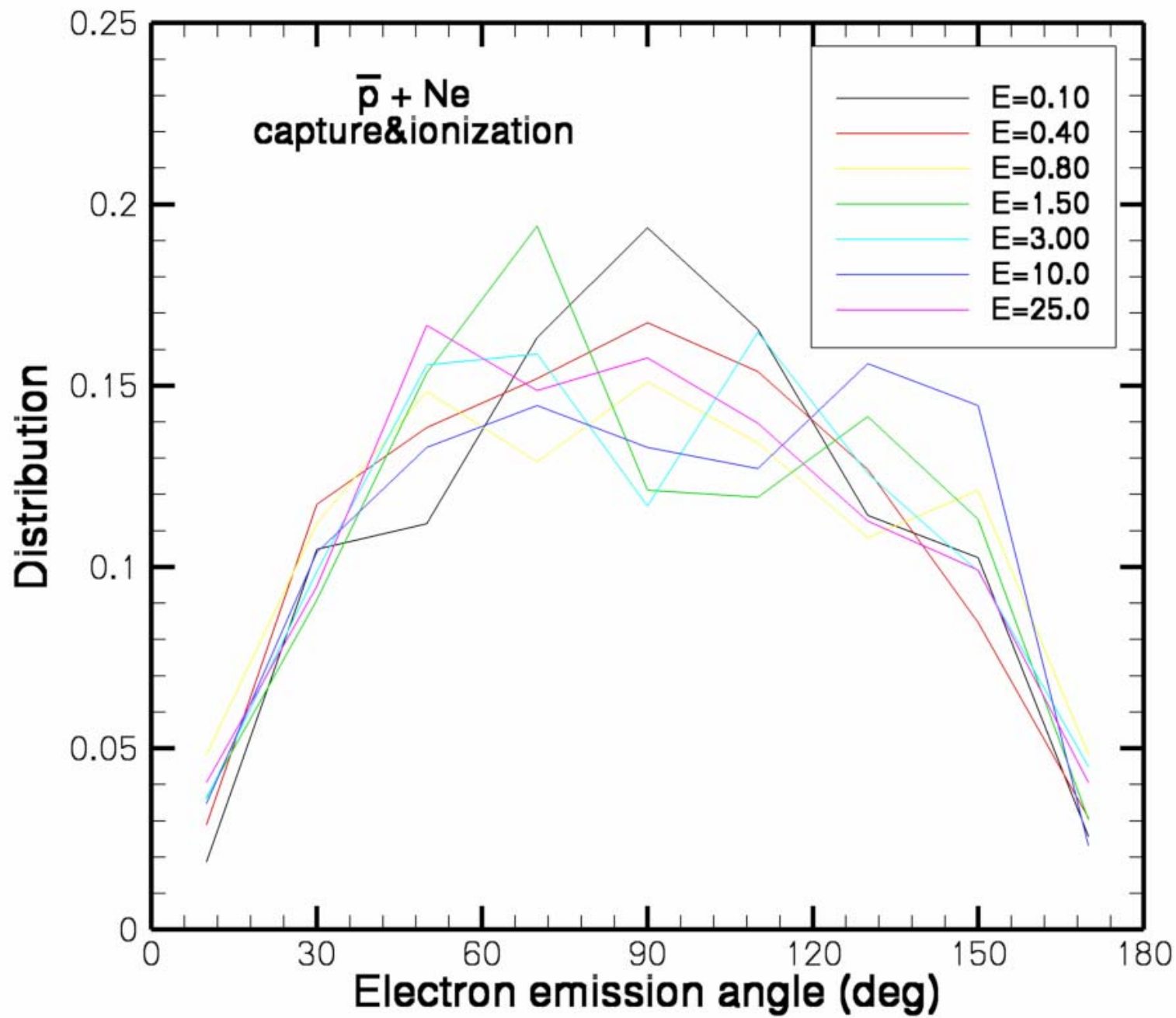
$$E_i^{(p)} - E_f^{(p)} > E_{\text{bind}}^{(e)}$$

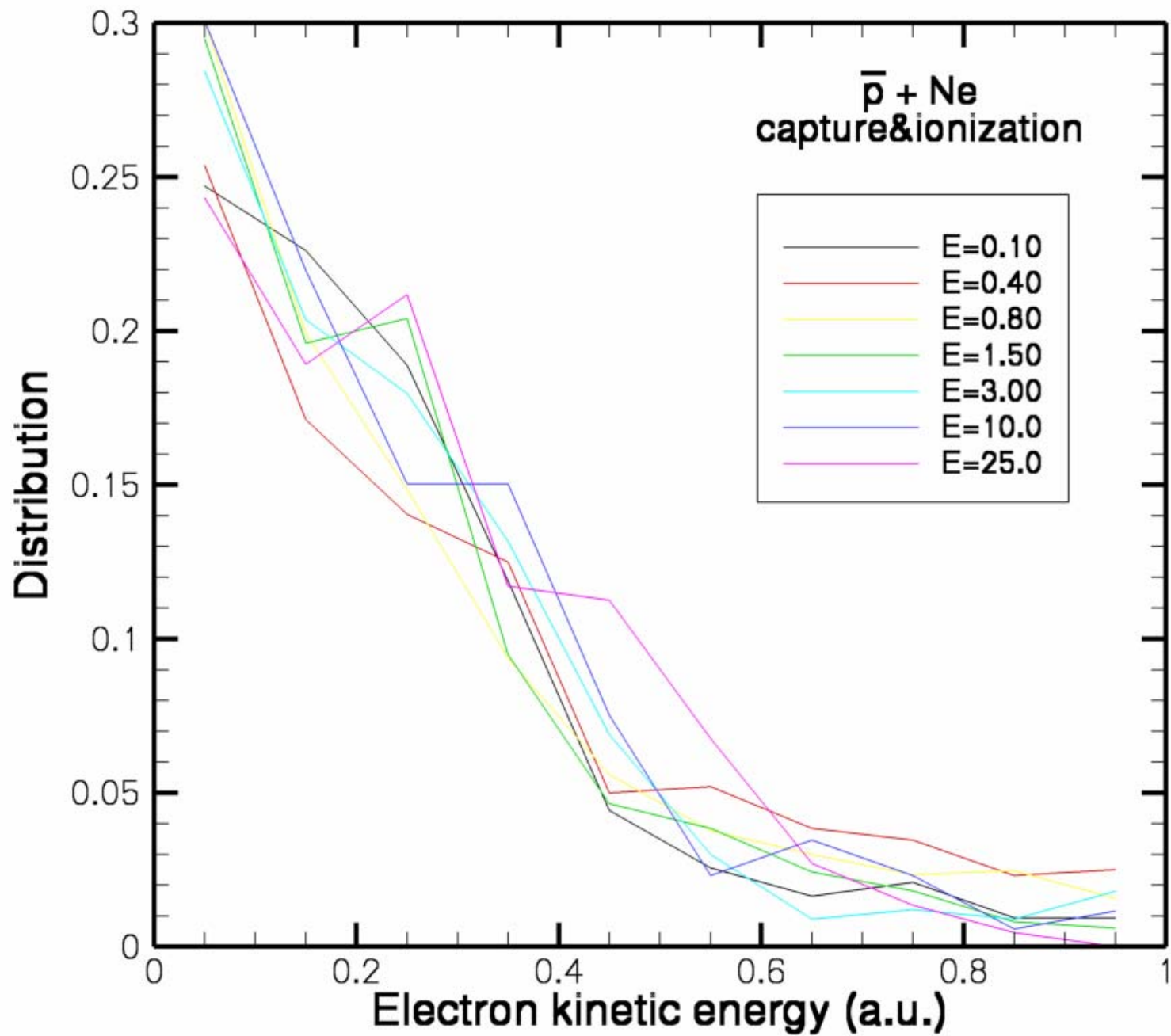
If $l=n-1$, then $\Delta n = 1$ for dipole transitions, and this condition is satisfied only for $n \lesssim 30$.

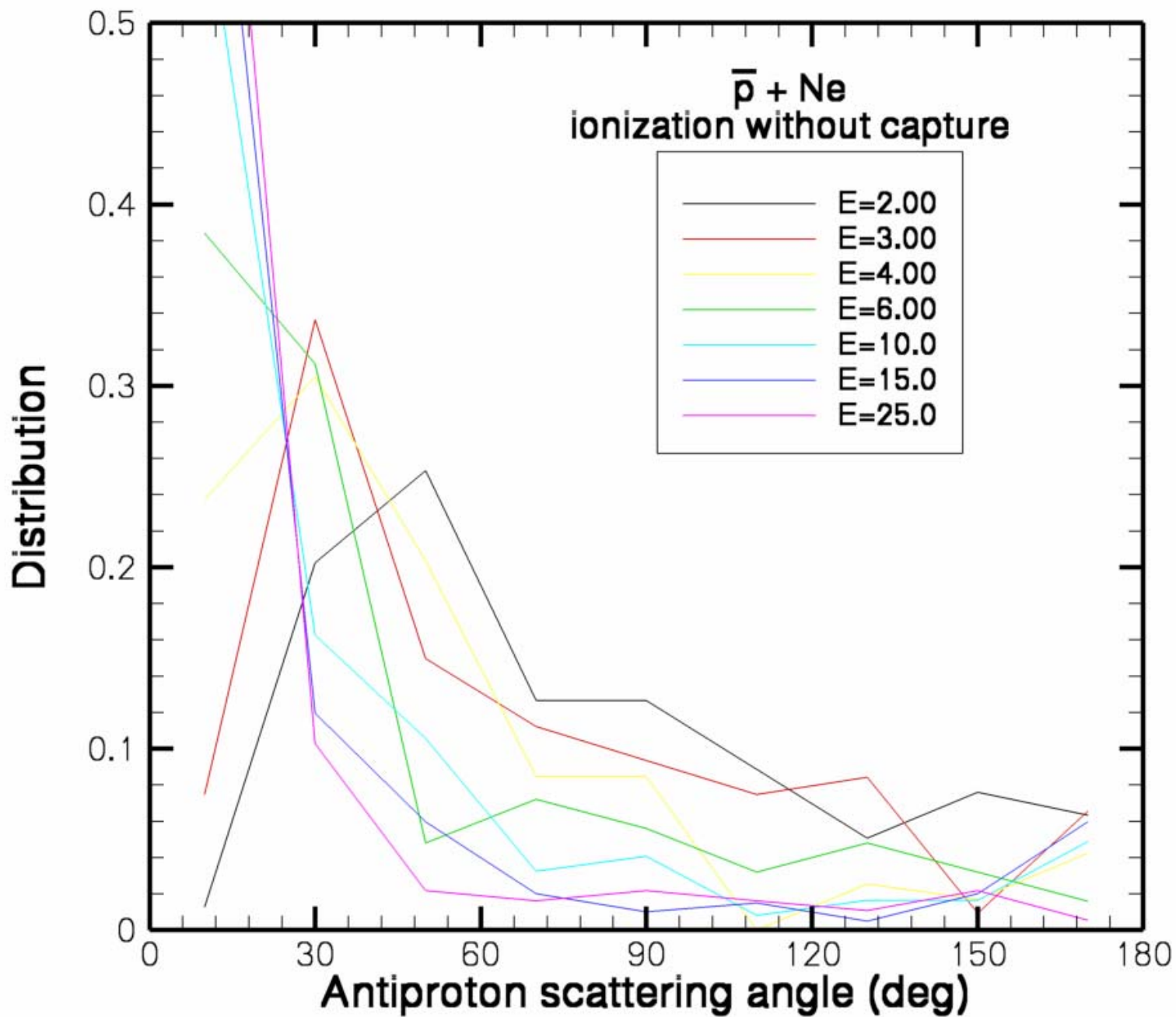
But big atoms (large n) may be collisionally **Stark mixed** so the selection rule doesn't apply.



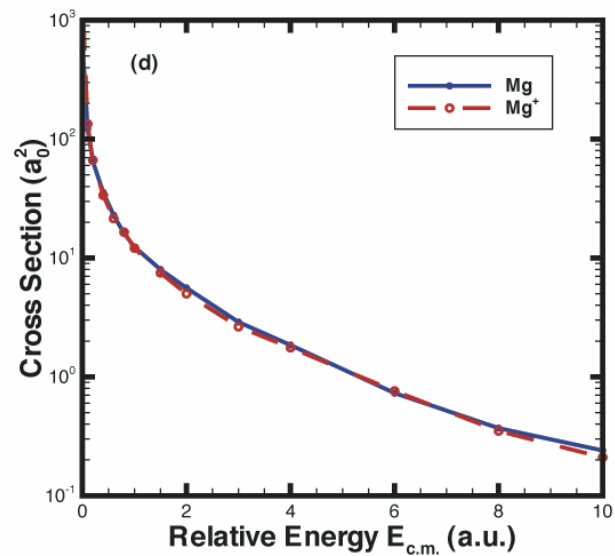
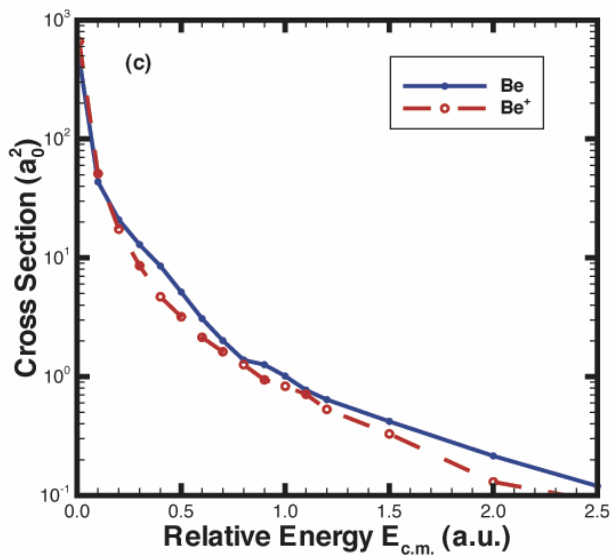
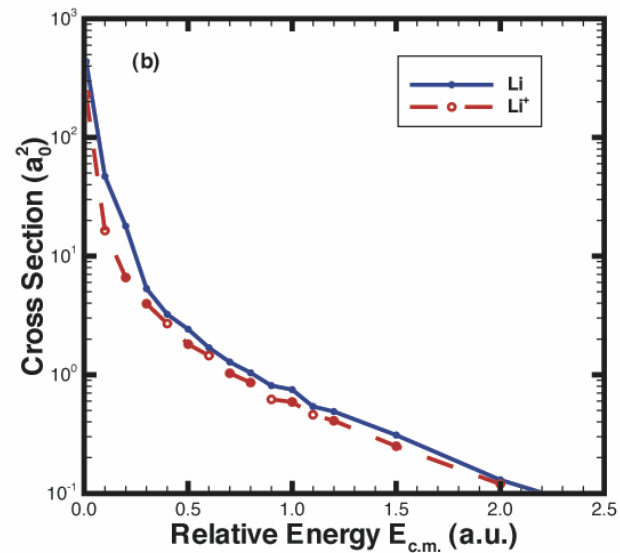
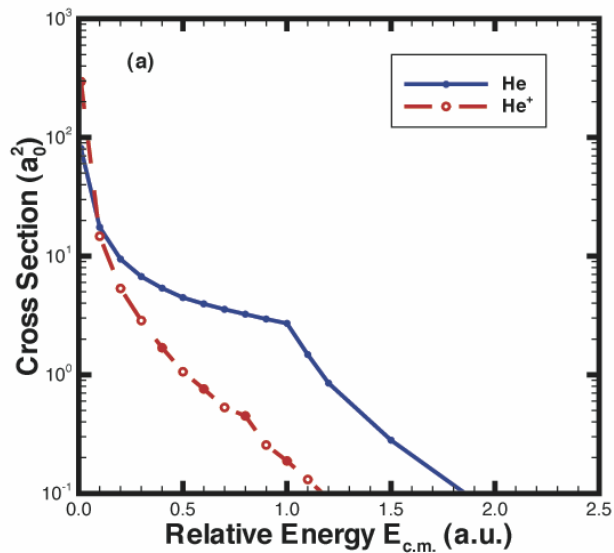




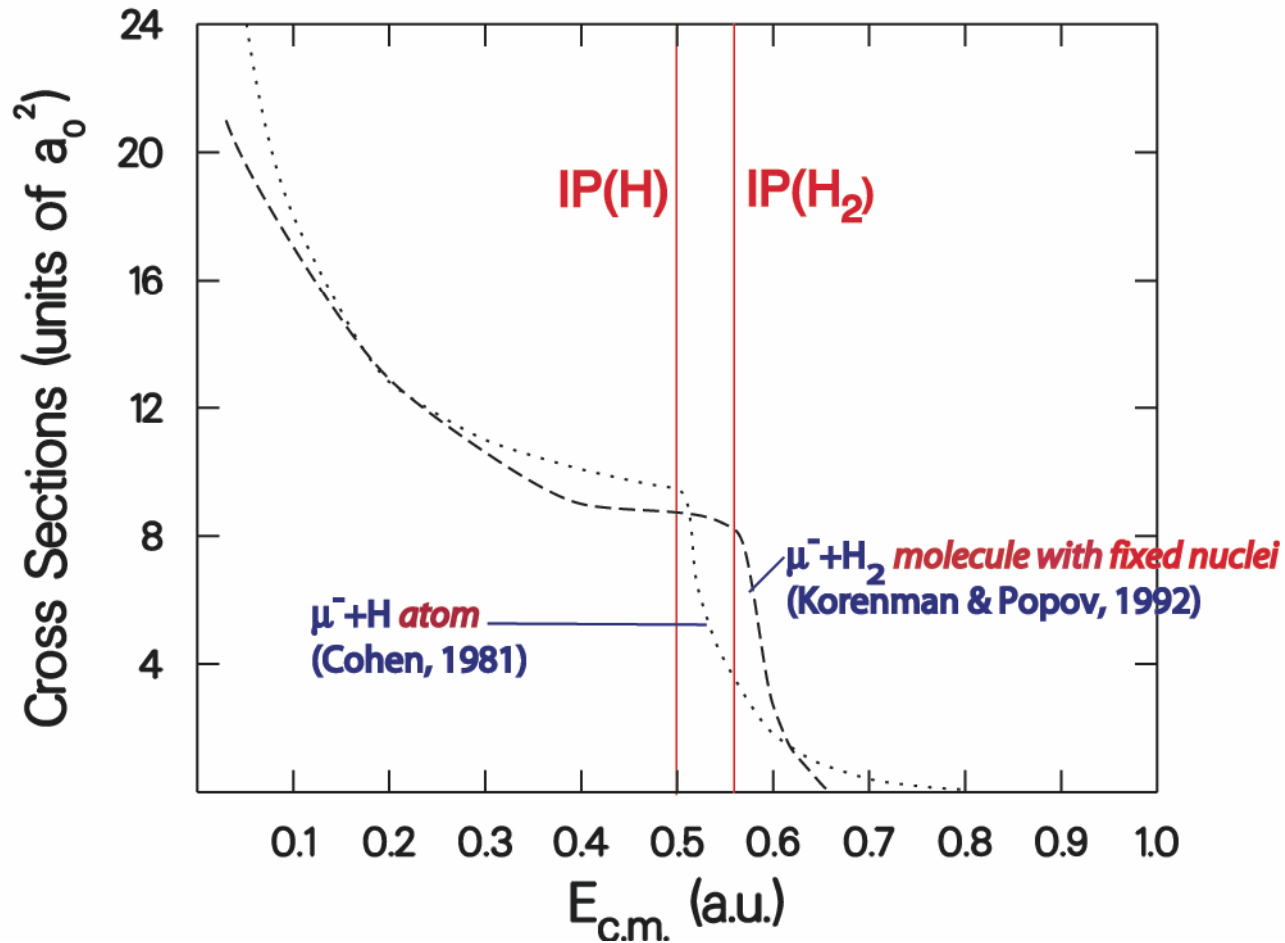


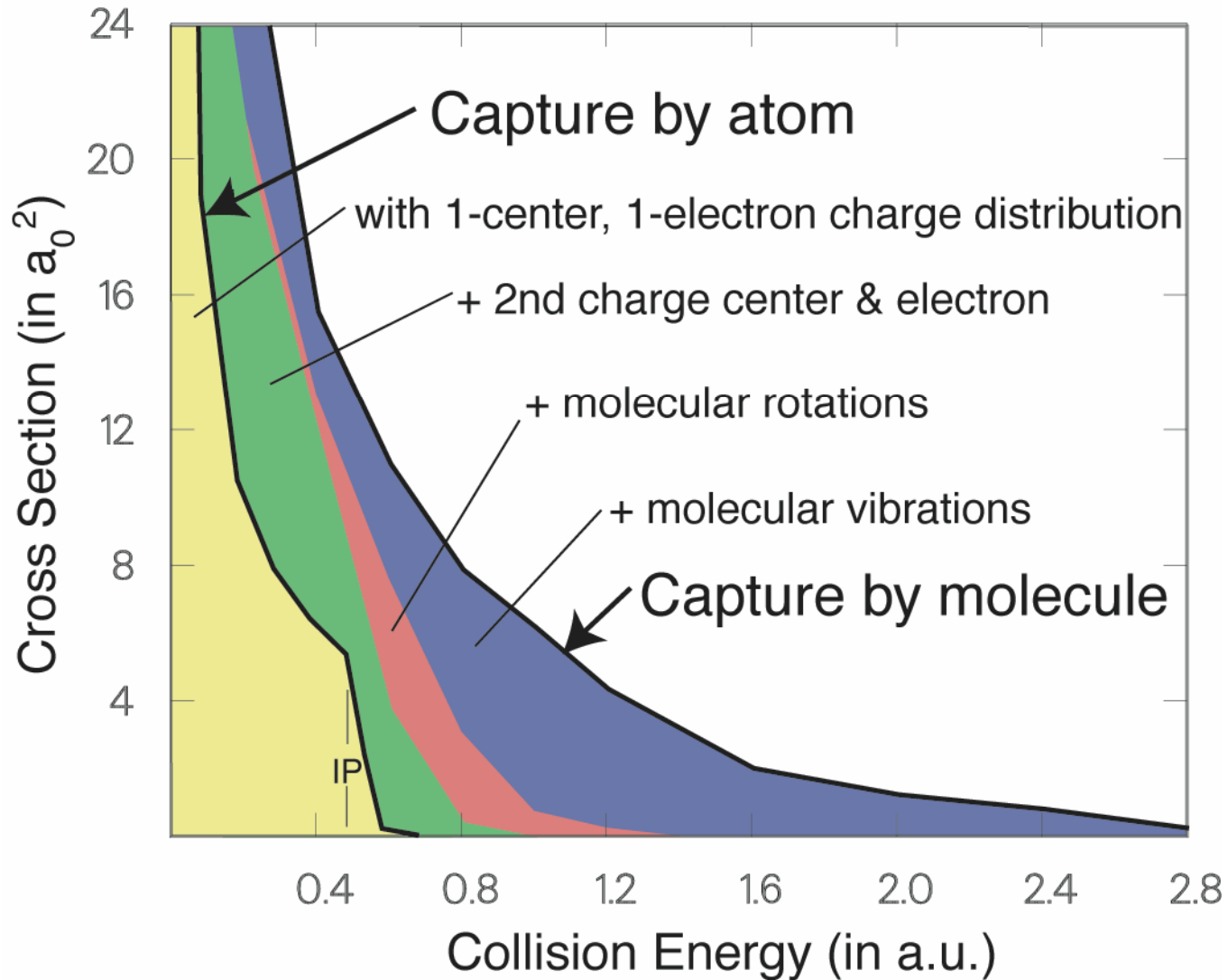
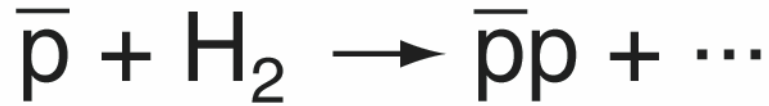


Capture of antiprotons by some (radioactive) atoms and their ions



Diabatic calculations
ignoring the internal structure of the molecule





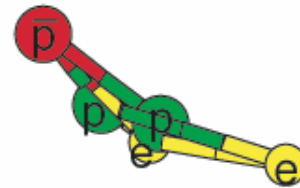
\bar{p} = antiproton p = proton e = electron

$\bar{p} \xrightarrow{22 \text{ eV}}$

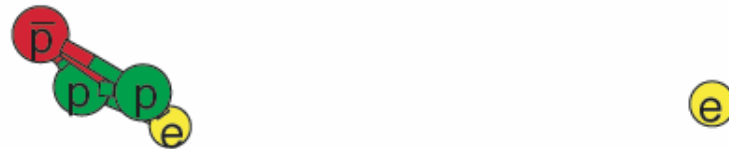
(a) $t=0$ (start)



(b) $t=246$ a.u.



(c) $t=259$ a.u.

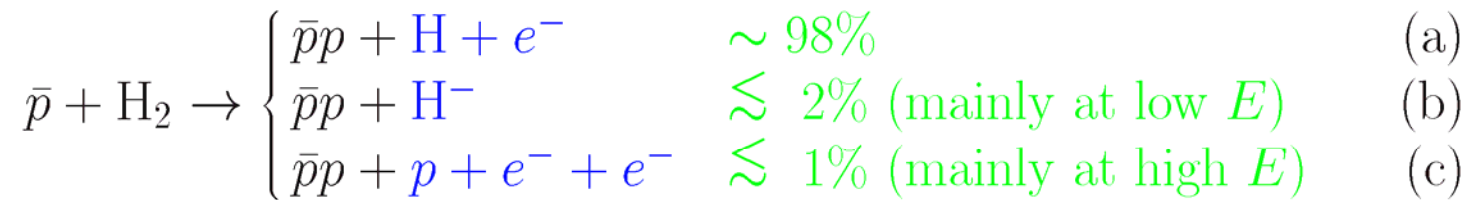


(d) $t=469$ a.u. (final state)

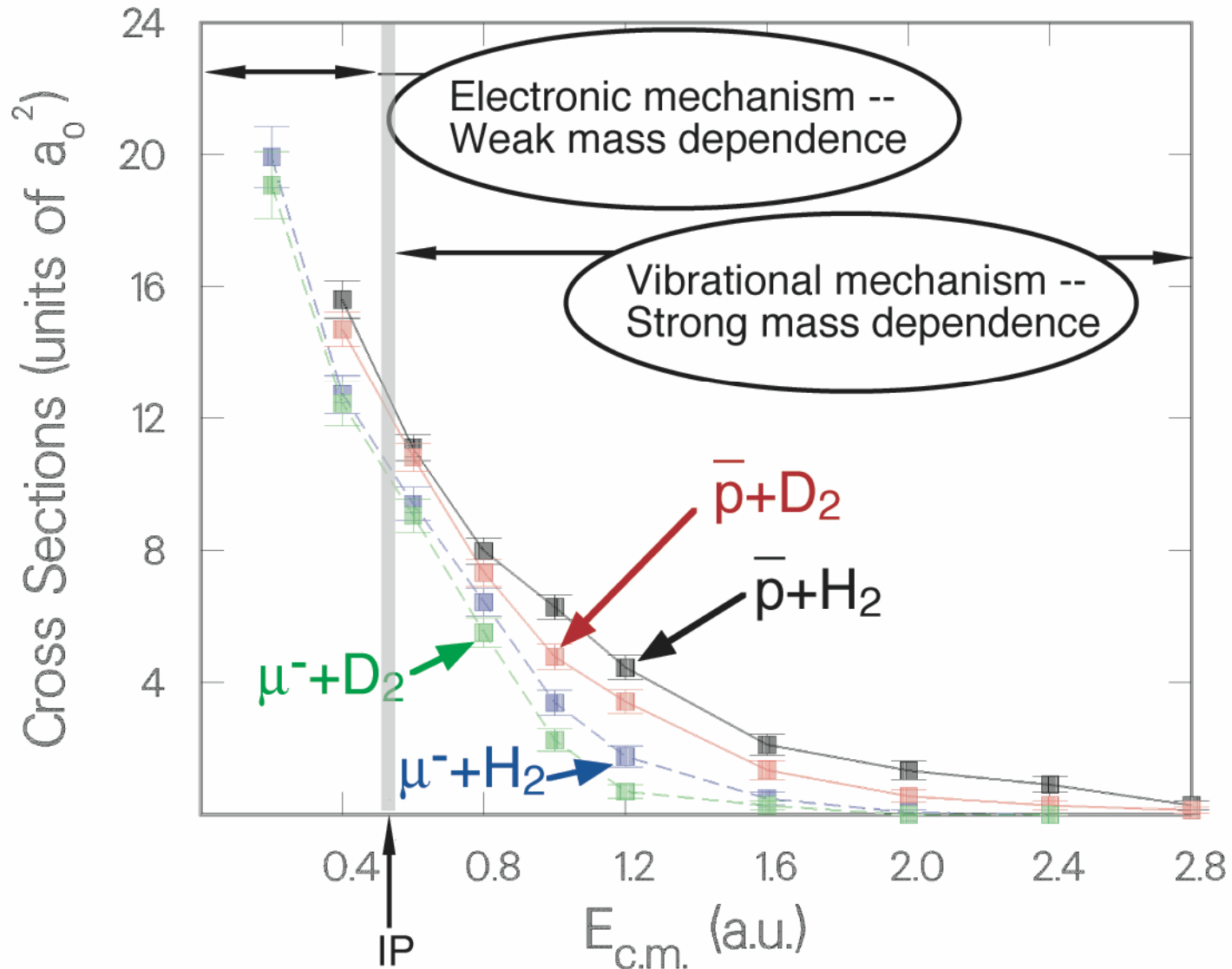


Final Arrangements in $\bar{p} + \text{H}_2$ Reaction

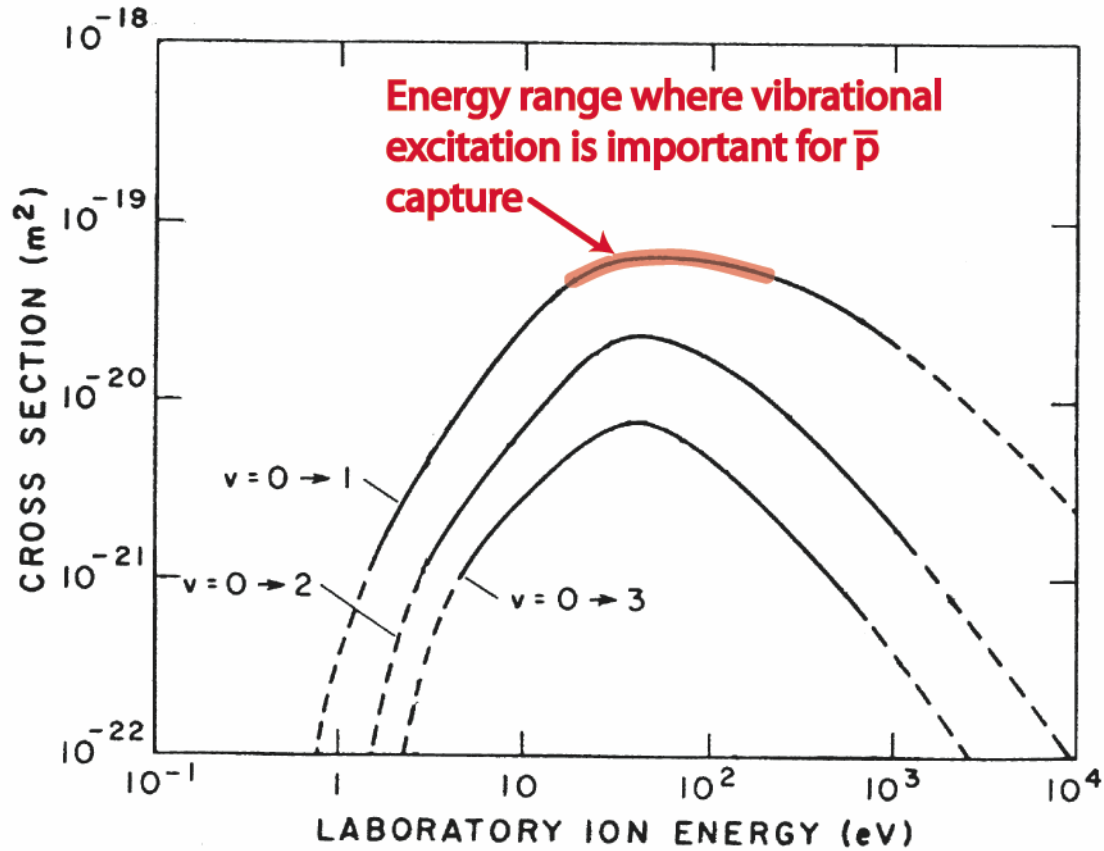
In capture of \bar{p} by H_2 , intermediate states such as $pp\bar{p}e^-$ or $p\bar{p}e^-$ are possible, but are **predissociative** and/or **autoionizing**. For this reason we have adopted the approach of following the trajectories long enough that the isolated $\bar{p}p$ atom can be characterized. The **reactions distinguished** are then



$\bar{p}p$, $\bar{p}d$, μ^-p & μ^-d Formation



Comparison with *nonexotic* physics



$$E_{lab} = 1.5 E_{c.m.}$$

Adapted from Figure 1 of A.V. Phelps,
J. Phys. Chem. Ref. Data **19**, 653 (1990).

Another comparison with *nonexotic* physics

"Lambert-Salter plot"

Collisional vibrational deactivation
of polyatomic molecules at 300K

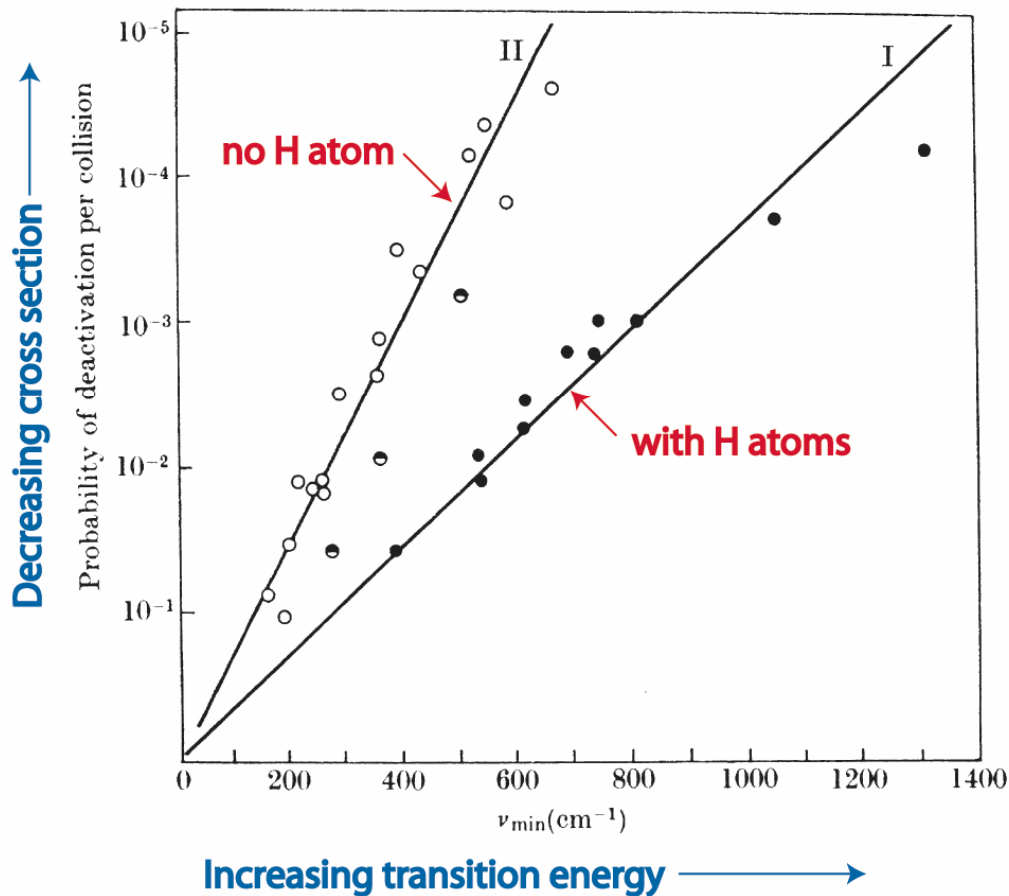


Figure 17.37 from Massey, *Electronic and Ionic Impact Phenomena*, vol. III

Capture of π^- in H_2+D_2 vs HD

P_p = Probability of initial capture by p ($P_d = 1 - P_p$)

W_p = Probability of final capture by p

$Q = (P_p - W_p)/P_p$ = Probability of transfer $\pi^- p + d \rightarrow p + \pi^- d$

Experimental: $W_p^{(\text{H}_2+\text{D}_2)} = 0.405 \pm 0.010$ [“World Fit”^{a(a-d)}]
 $W_p^{(\text{HD})} = 0.338 \pm 0.008$ ^b

Original experimental analyses *assumed* $P_p = P_d = 0.5$ in both H_2+D_2 and HD. With this assumption:

$$\left. \begin{array}{l} Q = (19 \pm 2)\% \text{ in } \text{H}_2+\text{D}_2 \\ Q = (32 \pm 2)\% \text{ in HD} \end{array} \right\} \text{Difference was a mystery!}$$

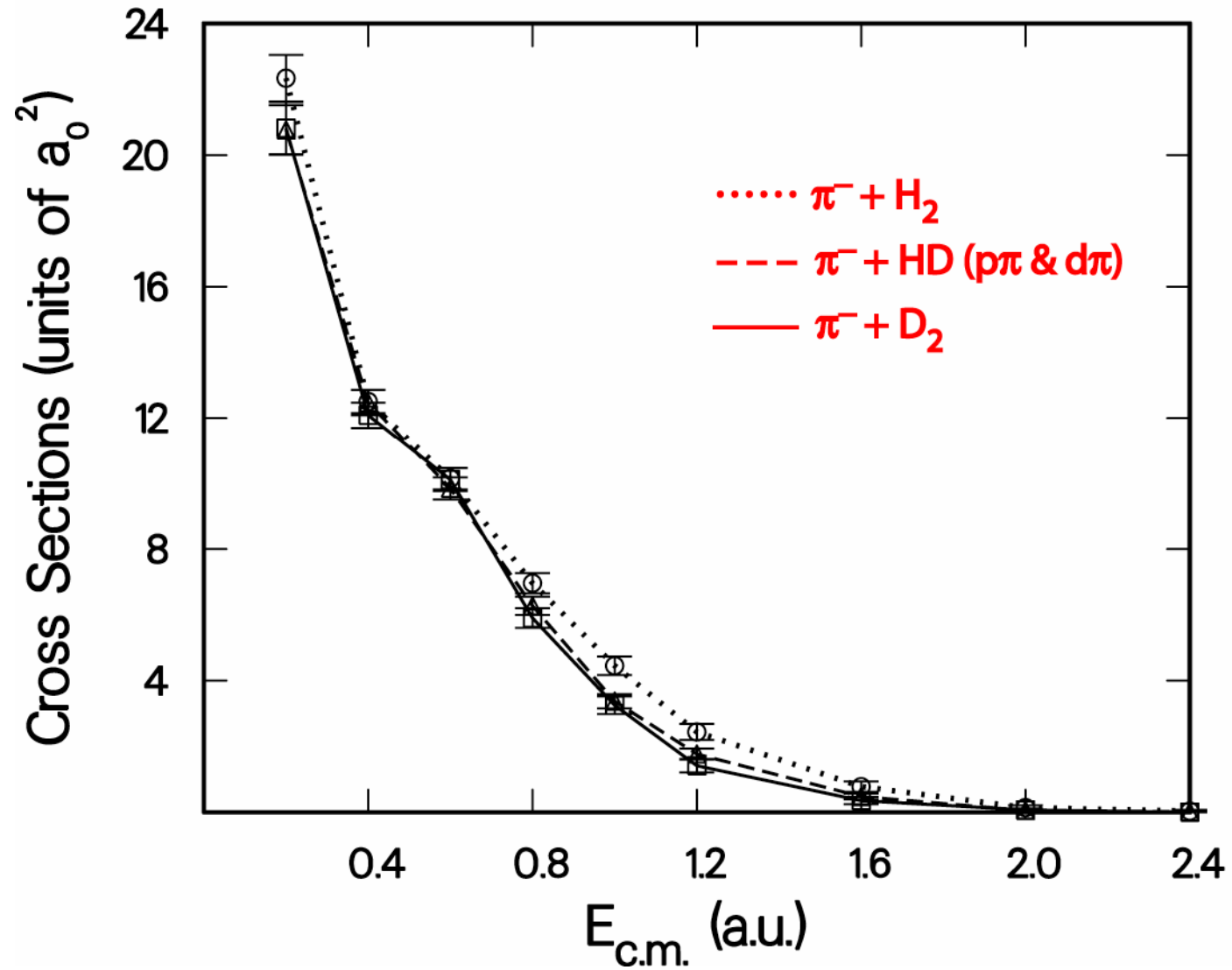
^aWeber *et al.* (1990)

^bAniol *et al.* (1983)

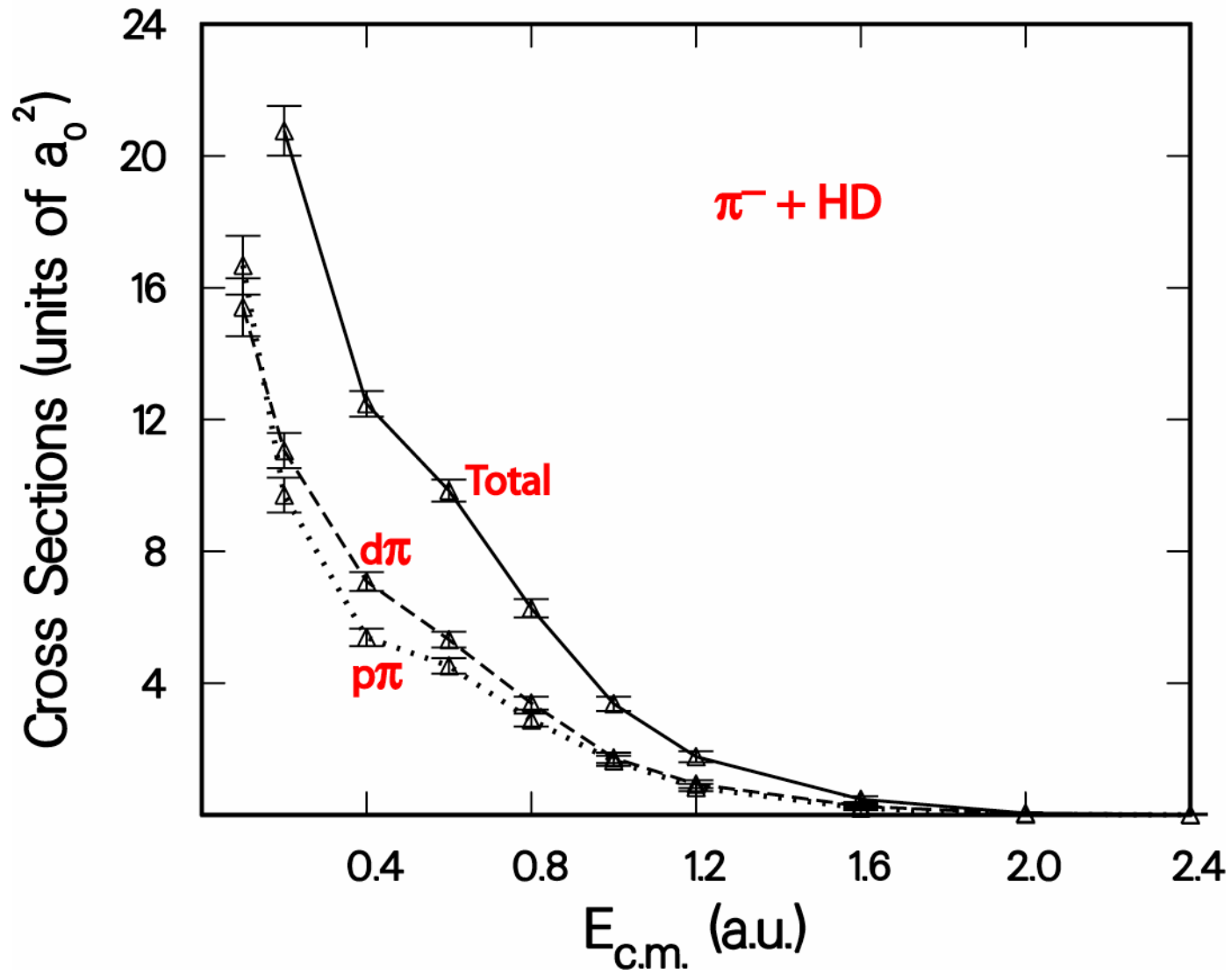
^cKravtsov *et al.* (1988)

^dPetrukhin & Prokoshkin (1969)

Negative pion capture by isotopic hydrogen molecules



Effect of dissociation



Capture of π^- in H_2+D_2 vs HD

P_p = Probability of initial capture by p ($P_d = 1 - P_p$)

W_p = Probability of final capture by p

$Q = (P_p - W_p)/P_p$ = Probability of transfer $\pi^- p + d \rightarrow p + \pi^- d$

Experimental: $W_p^{(H_2+D_2)} = 0.405 \pm 0.010$
 $W_p^{(HD)} = 0.338 \pm 0.008$

Original experimental analyses *assumed* $P_p/P_d = 1.0$ in both H_2+D_2 and HD. With this assumption:

$$\left. \begin{array}{l} Q = (19 \pm 2)\% \text{ in } H_2+D_2 \\ Q = (32 \pm 2)\% \text{ in HD} \end{array} \right\} \text{Difference was a mystery!}$$

Our FMD calculations give:

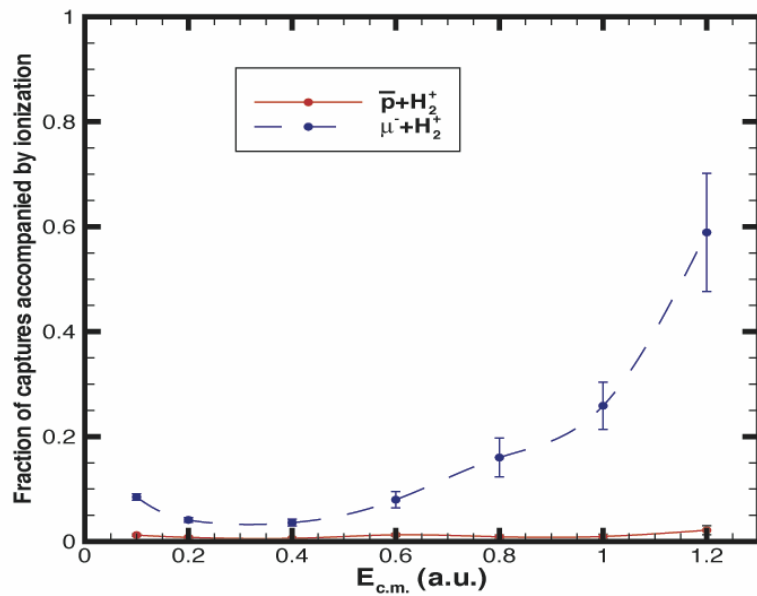
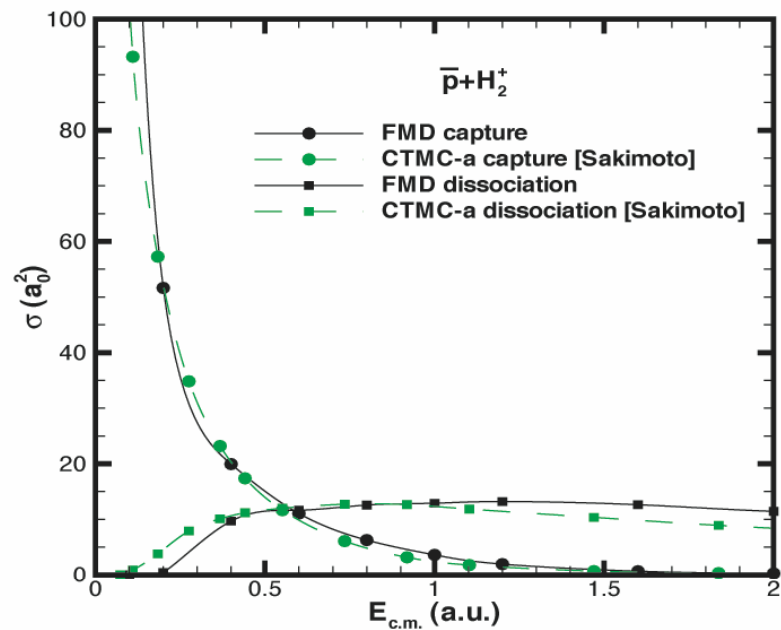
$$P_p/P_d = 1.204 \text{ in } H_2+D_2$$

$$P_p/P_d = 0.875 \text{ in HD}$$

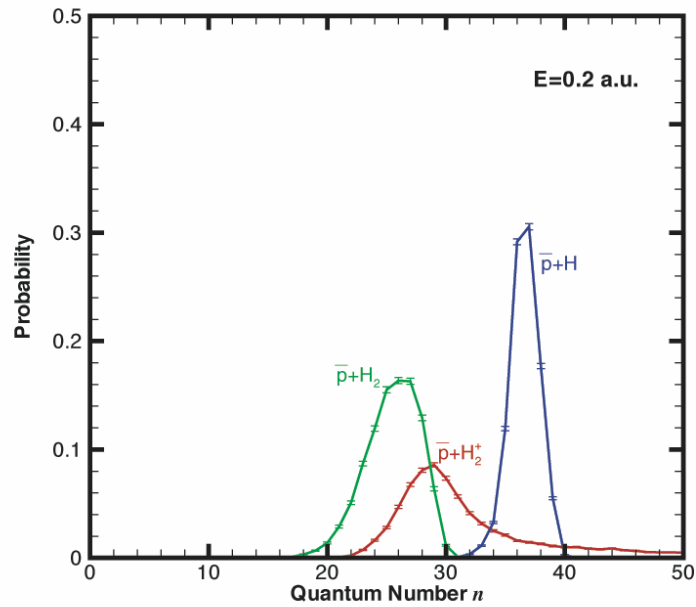
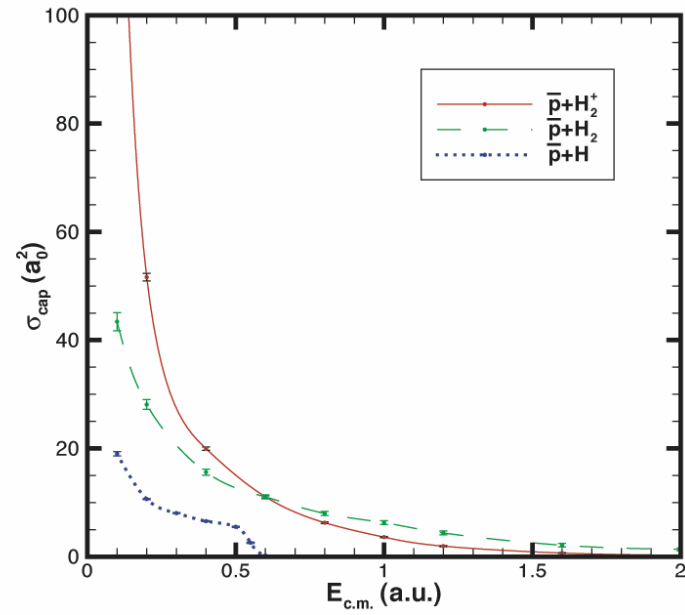
With these values of P_p/P_d , the *same experimental data* implies:

$$\left. \begin{array}{l} Q = (26 \pm 2)\% \text{ in } H_2+D_2 \\ Q = (28 \pm 2)\% \text{ in HD} \end{array} \right\} \text{Consistency is now quite satisfactory.}$$

Nonadiabaticity & ionization in \bar{p} capture by H_2^+



Capture of \bar{p} by H, H₂ & H₂⁺



Conclusions

For capture by atoms:

1. The adiabatic-ionization model is valid only for the H atom. No one-electron model can be adequate for $Z > 2$.
2. The isotope effect is quite small for atoms.
3. The initial angular momentum distribution increases more rapidly than $(2l + 1)$.

For capture by atomic ions:

1. Except for He^+ , capture is similar to the parent atom.

For capture by H_2 , D_2 , or HD:

1. Capture cross sections are larger and extend to higher collision energies for the molecular targets than for the corresponding atomic target. This effect is greater for \bar{p} than for μ^- (or π^-).
2. The isotope effect is large for H(or D)-containing molecules.
 - Initial capture favors the lighter nucleus.
 - Subsequent dissociation favors the heavier nucleus.
3. The n and l distributions are narrower and shifted to lower values relative to the distribution for the corresponding atom.

For capture by H_2^+ :

1. Capture by H_2^+ is almost entirely due to target dissociation, not ionization.

Key features for verification in \bar{p} capture experiments

For capture by atoms and atomic ions:

1. H and He atoms: Rather sharp cutoff at $E > \text{I.P.}$
2. Higher- Z atoms: Multiple ionization contributes to capture.
3. Atomic singly charged ions: Similar to parent atom, *except* for He.

For capture by hydrogen molecules and ions:

1. H_2 molecule: Capture at much higher energies than for H atom.
2. H_2^+ molecular ion: Capture occurs mostly without ionization.

Speculations: (based on interpretation of existing calculations)

1. Capture by molecules not containing H or D, *e.g.* O_2 , may be similar to that of the constituent atoms.
2. Hydrides may be *qualitatively* like H_2 . (H_2O might be interesting.)

Of great basic interest:

- Comparison of cross sections and quantum-number distributions for capture of \bar{p} by H, H_2 , and H_2^+ .